# **Treatment of Surfaces with Ions in a Glow Discharge**



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# How to Remove a Contamination Layer from a Surface?



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- chemical methods:
  - chemical etching
  - electro polishing
- physical methods:
  - grinding
  - polishing
  - sputtering





## definition:

to cause the atoms of a solid to be removed from the surface by bombardment with atoms in a discharge tube

#### advantages:

- no residues
- applicable even for complex geometries
- no toxic materials needed
- no waste

#### disadvantage:

vacuum needed

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# **Sputtering**





# **Sputtering**



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- simple reflection of the incident ion
- creation of secondary electrons
- sputtering of surface atoms
- implantation of the incident ions in a buried layer at the end of the range

#### **Interaction of Ions with Solid Matter**



#### Ratio between nuclear and electronic stopping power

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nuclear stopping of ions in solids

single displacement  $E_2 > E_d$ occurs  $E_1 > E_d$  and multiple displacements occur  $E_2 > E_d$  $E_2 < E_d$ phonon dissipation replacement  $E_1 < E_d$ ,  $E_2 > E_d$ collision and phonon and  $Z_1 = Z_2$ dissipation  $E_1 < E_d$ ,  $E_2 > E_d$  $Z_1$  becomes and  $Z_1 \neq Z_2$ interstitial atom

displacement energy: E<sub>d</sub> E<sub>0</sub>=E<sub>1</sub>+E<sub>2</sub>

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# **Binary Collision Approximation**



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scattering angle of the center of mass system



direction after the hit of the incident ion

direction after the hit of the target atom

energy of the target atom



**Interaction Potential** 

Coulomb Potential: 
$$V(r) = \frac{1}{4\pi\varepsilon_0} \frac{Q_1 Q_2}{r} = \frac{1}{4\pi\varepsilon_0} \frac{Z_1 Z_2 e^2}{r}$$

screening of the nucleus surrounding electrons taken into account :

$$V(r) = \frac{1}{4\pi\varepsilon_0} \frac{Z_1 Z_2 e^2}{r} \Phi(r)$$

#### $\Phi(r)$ screening function



#### The following contributions have to be considered for the potential:

- The classical terms of Coulomb interaction between
  - the nuclei of the particles
  - the electrons of the particles
  - the nuclei and the electrons of the other particle
- Quantum mechanical corrections due to the Pauli principle in the overlapping area and the interaction energy of electrons



#### universal screening potential:

 $\Phi(x) = 0.1818e^{-3.2x} + 0.5099e^{-0.9423x} + 0.2802e^{-0.4028x} + 0.02817e^{-0.2016x}$ 

## x dimensionless reduced radius

$$x = \frac{r}{a_I}$$

scaling length suggested by Ziegler, Biersack and Littmark, depending on the charges of the involved atoms

$$a_I = 0.8854 \frac{a_0}{Z_1^{0.23} + Z_2^{0.23}}$$

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 $[a_{I}] = 10^{-10} m$ 

## **Screening Function**



screening potential with radius for many atom pairs

screening potential with reduced radius for many atom pairs

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scattering angle of the center of mass system



direction after the hit of the incident ion

direction after the hit of the target atom

energy of the target atom



# **Probability of Displacing an Atom**

displacement energy threshold for Ni	$\Delta E = E_d = 17 eV$
mass Ni	M <sub>Ni</sub> =26.982u
mass Ar	M <sub>Ar</sub> =39.948u
energy of incoming ion	E <sub>0</sub> =150eV
sin <sup>2</sup> (Θ/2)	>0.228
impact parameter p	<1.14*10 <sup>-10</sup> m
lattice parameter a	3.52*10 <sup>-10</sup> m
probability of hitting an ion with more than the displacement energy threshold	33%

#### **Experimental Sputtering Yield**



Sputtering yields of nickel at 150eV ion energy vs argon gas pressure

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#### **Depth of Damage**



The depth of damage caused to tungsten following bombardment with low energy neon, argon and xenon ions.

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- material of the target
- type of gas
- pressure of gas
- electrical parameters
- temperature of the target



#### Pressure



Sputtering yields of nickel at 150eV ion energy vs argon gas pressure

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#### sputtering rate R



#### Measuring R is difficult because:

- removal of only a few layers
- variations of the gas pressure
- the composition of the gas
- the incident ion flux
- problems due to difficulties in measuring the depth
- compositional changes of the target
- degradation

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# **Compositional Changes of the Target**



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#### sputtering yield Y

- unit: atoms/ ion or grams/ion
- assumption for the calculation of N<sub>i</sub>: one argon ion is generated by one electron
- assumption: pure material without contamination

 $Y = \frac{N_s}{N_i} = \frac{\text{number of removed atoms from the target}}{\text{number of injected ions}}$ 

$$N_i = \frac{I \cdot t}{Q_{e^-}} = \frac{\text{current} \cdot \text{time}}{\text{charge of } e^-}$$

 $N_{S} = \frac{\Delta m \cdot N_{A}}{M_{mol}} = \frac{\text{change of mass} \cdot \text{Avogadro constant}}{\text{molar mass}}$ 



#### **Sputtering Yield**



Sputtering yield of MgO by a) He and b) Ar ion beams. The solid lines correspond to the fitted function to the measured data.

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#### **Sputtered Aluminum Samples**



SEM micrographs of sputtered aluminum covered with an oxide layer (0.2mbar Ar, 400°C) after a) 30min, b) 45min, c) and d) 60min

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# THE DEVIL IS IN THE DETAILS

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